



Potash lime silica glass from renewable raw materials in the system rice husk ash - eggshells - beech wood ash

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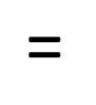
Motivation









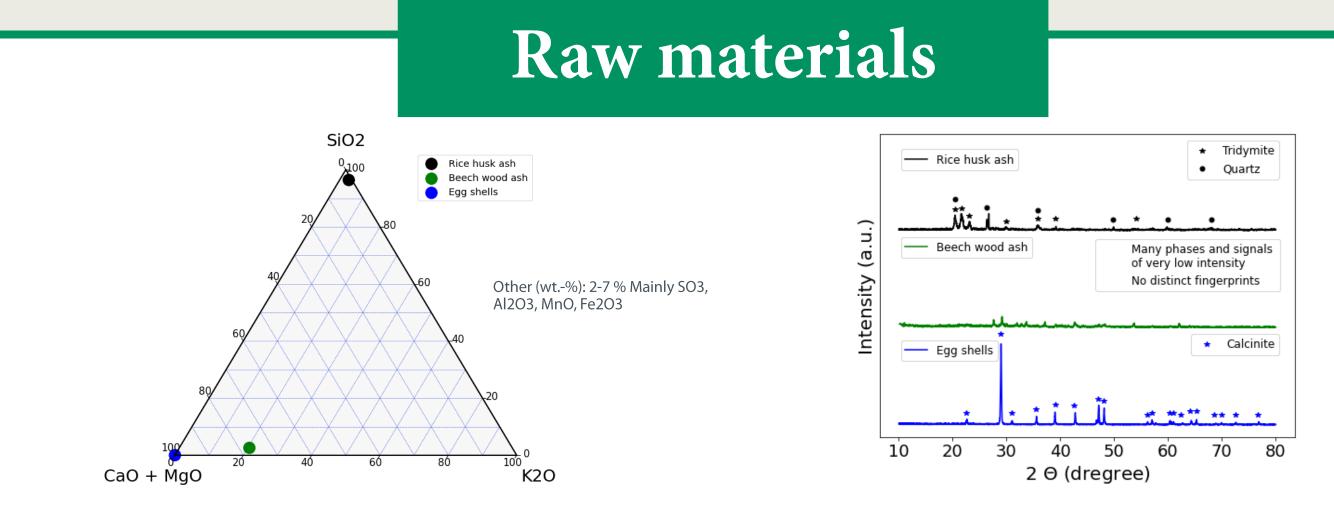




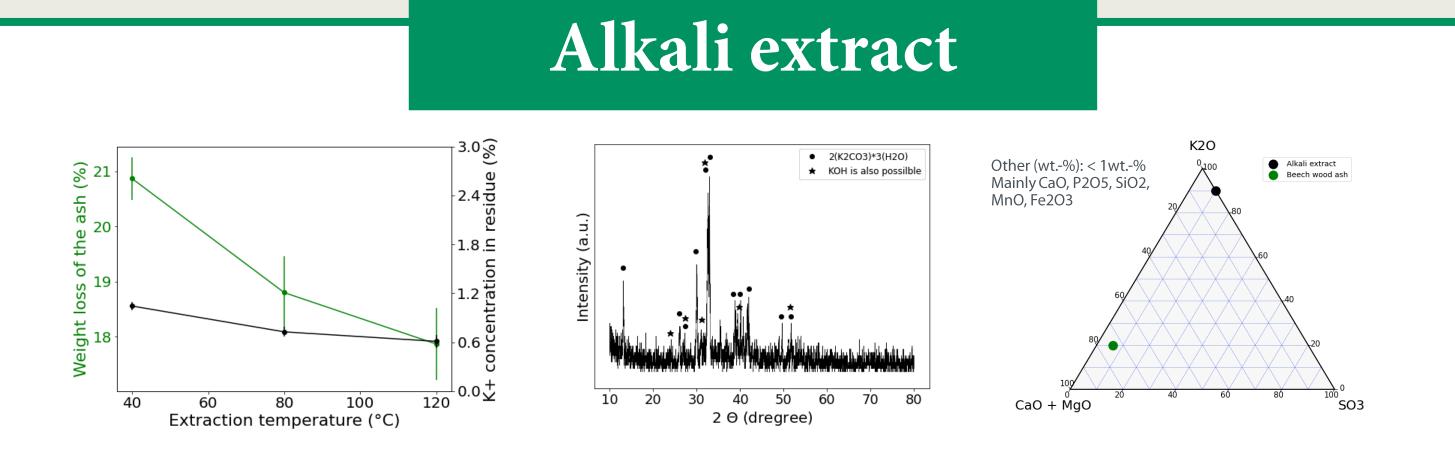
- Food production generates a large amount of agricultural byproducts [1]
- These wastes can also be used for glass production [2]
- Compared to synthetic raw materials, there are challenges [3]
 - Impurities and undesirable components
 - Alkali content of ashes is usually limited to 20-40 %
 - Ashes usually contain only K₂O as alkali and very rarely Na₂O
- The overall aim of the work is to produce glass from exclusively renewable raw materials with properties close to those of container glass

Approach

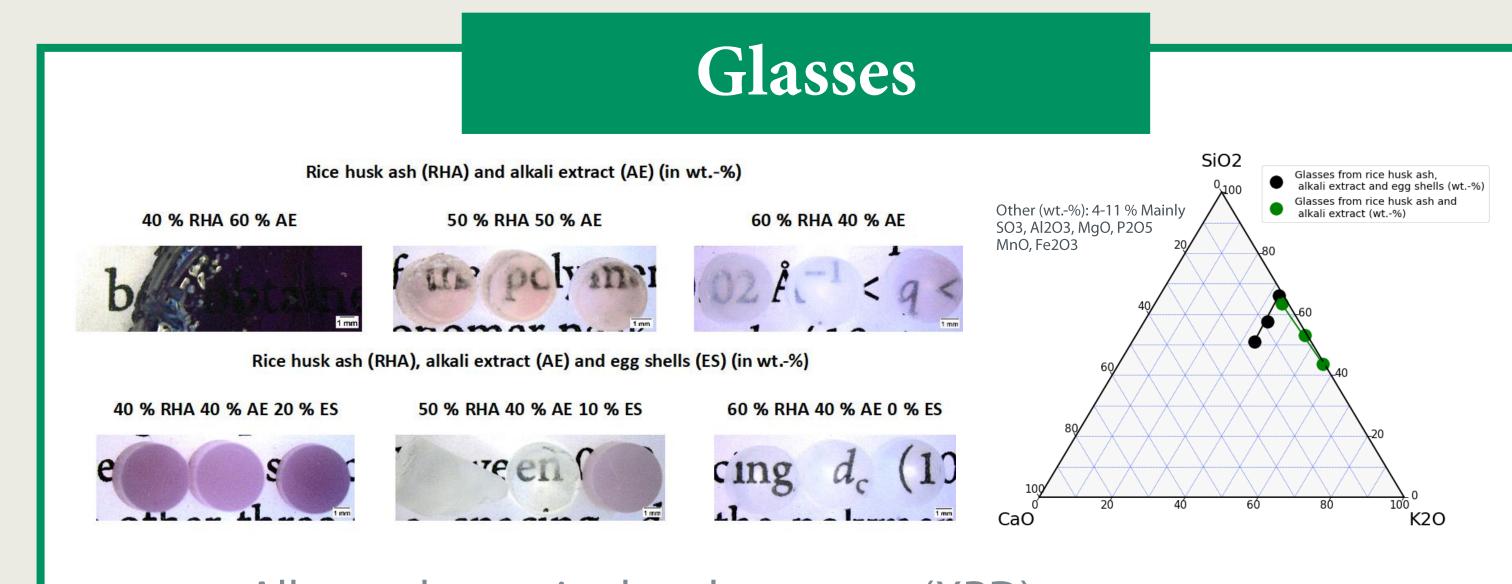
- The raw materials selected were
 - Rice husk ash (RHA) as a SiO₂ source
 - Beech wood ash as an alkali source (BWA)
 - Egg shells (ES) as a CaO source
- The alkalis from the beech wood ash were extracted with water for 4 h and the alkali extract (AE) was dried
- SciGlass® was used to calculate the Littleton point. The glasses from the batch compositions are within the Littleton point range of < 780 °C
- Two series of batch compositions were selected
 - (x) RHA (100-x) AE Batch compositions are in wt.-%
 - (x) RHA (40) AE (60-x) ES with x = 40, 50, 60 %
- Melting temperature: 1450 °C with 1 h dwell time, cooled down in the furnace



- Rice husk ash and eggshells contain mostly SiO₂ and CaO
- Beech wood ash is relatively low in alkali content



- 40 °C is sufficient to extract alkalis from beech wood ash
- The dried alkali extract contains mainly potassium compounds



- All samples are in the glassy state (XRD)
- Glasses from alkali extract and rice husk are transparent and colored (MnO), the ones with eggshells got opaque

Summary

- Rice husk ash and egg shells are relatively pure raw materials which can be adequately used as resources for SiO₂ and CaO
- The beech wood ash as a source of alkali contains relatively few alkalis so a chemical extraction is necessary
 - Alkalis are easily extracted with water at low temperature and the extract contains mostly potassium compounds
- All the samples are in the glassy state, but mostly colored due to presence of Mn oxide, and glasses without egg shells are transparent